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**The unit cell and space group of *N*-acetyl-L-tyrosylamide.\*** By G. B. CARPENTER. *Gates and Crellin Laboratories of Chemistry, California Institute of Technology, Pasadena 4, California, U.S.A.*

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The program of work in these laboratories on the structure of amino-acids and their derivatives includes a study of the structure of tyrosine. Since we had not obtained good crystals of tyrosine itself, we sought a crystalline derivative of tyrosine, preferably one in which the substituents would be similar to the immediate neighbors of a tyrosine residue in a polypeptide chain. To this end we examined crystals obtained by the crystallization of racemic *N*-acetyltyrosylamide to see if they might be suitable for use in a complete X-ray diffraction determination of the structure of tyrosine.

Clear crystals† of racemic *N*-acetyltyrosylamide and of *N*-acetyl-L-tyrosylamide were very similar and appeared to be tetragonal. Crystals of the inactive form were examined first in the hope that they might contain a center of symmetry. Laue photographs revealed that they are indeed tetragonal with a Laue symmetry of  $D_{4h}$ . Weissenberg photographs showed  $h00$  present only for  $h$  even and  $00l$  present only for  $l$  even; the absences are characteristic of the space group  $P4_22_1-D_4^2$ . This space group does not contain a center of symmetry; the general positions are eightfold.

Layer-line measurements on rotation photographs taken with a camera of 5 cm. radius using Ni-filtered  $Cu K\alpha$

radiation ( $\lambda = 1.5418$  A.) gave the following axial lengths:

$$a_0 = 10.84 \pm 0.04, \quad c_0 = 20.31 \pm 0.05 \text{ A.}$$

Assuming eight molecules per unit cell leads to an X-ray density of  $1.24 \text{ g.cm.}^{-3}$ , which is not far from an experimental density of  $1.19 \text{ g.cm.}^{-3}$  measured by the flotation method.

The space group could accommodate both *D* and *L* molecules in the unit cell only if there were at least sixteen of them, i.e. only if the asymmetric unit contained both a *D* and an *L* molecule; therefore it appeared that the racemate must resolve spontaneously on crystallization. This conclusion was verified by Laue photographs of *N*-acetyl-L-tyrosylamide, which proved to be identical with the corresponding photographs of the crystals from the racemate. The densities also appeared to be identical. Further, the melting-point of single crystals isolated from the racemate was found to be the same ( $224-226^\circ \text{C.}$ ) as that of the *L* material, whereas a powder of numerous crystals from the racemate melted around  $198^\circ \text{C.}$  (These melting-points were determined by Mr Robert MacAllister; they are corrected.)

Because of the large unit cell and the absence of a center of symmetry, an exhaustive determination of the structure of *N*-acetyl-L-tyrosylamide does not at present appear sufficiently profitable.

We wish to express our thanks to Dr Robert B. Corey who suggested this investigation.

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† Prepared by Mr Robert MacAllister of these laboratories.

## Notes and News

*Announcements and other items of crystallographic interest will be published under this heading at the discretion of the Editorial Board. Copy should be sent direct to the British Co-editor (R. C. Evans, Crystallographic Laboratory, Cavendish Laboratory, Cambridge, England).*

### Commission on Crystallographic Nomenclature

Members of the Commission on Crystallographic Nomenclature of the International Union of Crystallography (see *Acta Cryst.* (1948), 1, 341) have now been nominated as follows:

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Crystallographers interested in the work of this Commission are cordially invited to make contact with the Chairman.

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Notice of adhesion in Group IV, dated 2 June 1949, has been received from Spain through the Consejo Superior de Investigaciones Cientificas.